PURCHASE ORDER

BEKER INDUSTRIES CORP.

P. O. BOX 37, CONDA, IDAHO 83230 (208) 547-4381

то;		198	ff Industries, Inc. 2 West Jefferson Ave. roit, Michigan 48206	INSTRUC 1: To val promp 2: List Ord Shippin	Page: 1 0 TIONS TO VENDO Idate this order, retur tly duly signed der No. and Item No. a g Document:	3-401A j 2 R n acknowled in all involces.	Pockages an
		Beker Industries Corp. P. O. Box 37 Conda, Idaho 83230 Why se specified 2% discount will be deducted for payment 10th proximo: 11st Ext		3: If shipped unit quantities are less than ordered contend purchasing Manager whose signature appears below. 4: Describe in detail on invoice item(s) shipped and Quantitie 5: Send invoices to: Beker Industries Corp. Co Accounts Payable P. O. Box 37 Conday, Idaho 83230			s below.
PROMISED			F:O.B.POINT	SHIP VIA	TR FREIGHT	IG.TERMS:	FREIGHT PPD COL
ITEM NO.	QUANTITY	U/M	MATERIAL AND DESCRIPTION		UNIT PRICE	TOTAL	AMOUNT
		ion Sta equ Gyp dat dra	ff industries, Inc. is to provid- igment and materials to line the sum Pond in accordance with their ed July 11, 1975, and in accordance wings provided by Beker Industri	e labor, No. 3 r quotati nce with es Corp.	ion		
		ant ezc. pzo the cuo	Contractor is to provide a one- ee on materials and labor subject lusions noted in the quotation. vide, further, a five-year guara- sidewall material as specified tation, bottom liner material is to be the meets the standard specificat.	t to the He is to ntee on In this			
		pol	yvinyl chloride plastic linings a Staff Industries. BEKER INDUS	as snowi			
and condition	y accepting the printed on	the reverse	grees to the general terms	f Purchasing: roperly which I he e: Primarily to the g or for processi	nverburckised Remodings process of producing language timery described above who cing and on of	ible personal pr	operly by mini clly and primar

Purchase Order No. COF

Date:

BIC P8 9M 4-75

PURCHASE ORDER CONTINUATION SHEET

BEKER INDUSTRIES CORP.

Purchase Order No. COF 20841

Date: July 22

Requisition No. 635
Cost Code No. 168-401A

Page: 2 of 2

ITEM NO.	QUANTITY	U/M	MATERIAL AND DESCRIPTION	UNIT PRICE	TOTAL AMOUNT
			The sidewall material is to be polyester reinforced coated FVC 8422 as manufactured by Shelter-Rite Corporation:		
			The Contractor is to provide adequate liab- lility coverage for his employees and is to provide Beker Industries Corp. with evidence of such coverage.		
			The Contractor is to apply such cover materias, in his judgment, may be required to secuthe liner until a water cover is established he will also provide such personnel as may be required to remove any air pockets which may develop as the initial pond bottom cover is established.	re I	
			The Contractor shall make all reasonable eff to complete the installation in an expeditio and workmanlike manner.	Xt Is	
			Beker Industries Corp. will provide a graded surface on the pond bottom, grade the pond w to a degree suitable for sidewall liner inst lation, and will provide one access road to the inside of the pond. Beker Industries Corp. will also remove the access road at the appropriate time as indicated by the Contrac	115 11-	
		12.2	Lump sum payment will be made by Beker Industries Corp. after acceptance of the complete installation.		
			Sum of:		\$258,491.00
	Mistria i				

RIC 84 5M 4-75

Beker Industries Corporation CONDA, IDAHO 83236

635

PURCHASE REQUISITION

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Beker Industries Corporation

PURCHASE REQUISITION

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		<u> </u>		VENDOR'S SHIPPING DATE	BY		

STATEMENT

Our New Phone No. is: (313) 496-0808 Iolophone (313) 259-1820

THIS STATEMENT SHOWS OPEN ITEMS AS OF THIS DATE



INDUSTRIES INC.

'est Jefferson Avenue • Detroit, Michigan 48216

Beker Industries, Corp P.O. Box 37 Conda, Idaho 83230

JRN THIS STUB WITH YOUR CHECK

AMOUNT	•	

<u> </u>	INVOICE NO.	CHARGES	CREDITS	BALANCE
	085-102			\$30,289.50
	085-101			43,757.97
		SEP 8 10/5		
				TOTAL DUE
	1	,		\$74,047.47

ITS ARE DUE AND PAYABLE IN ACCORDANCE TO TERMS STATED ON INVOICE



Telephone (313) 259-1820

NDUSTRIES INC. 1982 West Jefferson Avenue • Detroit, Michigan 48216

FH0028586

STAFF INDUSTRIES, INC. 78 Dryden Road Upper Montclair, N. J. 07043 201-744-5367

STANDARD SPECIFICATIONS

POLYVINYL CHLORIDE PLASTIC LININGS

O1 - GENERAL REQUIREMENTS

The work covered by these specifications consists of installing a polyvinyl chloride (PVC) plastic lining in the reservoir and/or canal where shown on the drawings or directed by the Engineer. All work shall be done in strict accordance with the drawings and these specifications and subject to the terms and conditions of the contract.

02 - PVC MATERIALS

- A. General. The materials supplied under these specifications shall be first quality products designed and manufactured specifically for the purposes of this work, and which have been satisfactorily demonstrated by prior use to be suitable and durable for such purposes.
- B. Description of FVC Materials. FVC polyvinyl chloride plastic lining shall consist of widths of calendered polyvinyl chloride sheeting fabricated into large sections by STAFF INDUSTRIES, INC. by means of special factory-bonded seams into a single piece, or into the minimum number of large pieces required to fit the facility.
 - 1. Physical Characteristics. The PVC materials shall have the following physical characteristics.

TEST	TYPICAL TEST VALUES	TEST METHOD
Specific Gravity	1.24 - 1.30	
Tensile Strength, psi, min.	2200	ASTM D882-B
Elongation, % min.	300%	ASTM D882-B
100% Modulus, psi	1000 - 1600	ASTM D882-B
Elmendorfer Tear, gms/mil, min.	160	asim 689
Graves Tear, lbs/in. min.	270	ASTM DLOO4
Water Extraction, % max.	0.35	ASTM D1239
Volatility, % max.	0.7	ASTM D1203
Impact Cold Crack, F.	-20	ASTM 1790
Dimensional Stability, max. %		
(100°C-15 minutes)	5	
Shore Durameter, "A"	65 - 75	ASTM D676
Outdoor Exposure, sun hrs.	1500	mm bolo
Bonded Seam Strength, % of Tensile, min.	80%	
Pinholes/10 sq. yds. max.	1	
Resistance to Burial	•	Meets USBR Test
		(specially formulated
		for resistance to
•		micro-biological
		attack)
Alkali Resistances		Passes Corps of Eng.
		CRD-572-61
Color	Black or Grey	···· / · · ·

STAFF INDUSTRIES, INC.

2. PVC Polyvinyl Chloride Materials shall be manufactured from domestic virgin polyvinyl chloride resin and specifically compounded for use in hydraulic facilities. It shall be neutral gray to black in color and produced in a standard minimum width of 54 inches. Thickness shall be as shown on the project drawings.

03 - FACTORY FABRICATION

Individual widths of PVC materials shall be fabricated into large sections by solvent bonding by STAFF INDUSTRIES, INC. into a single piece, or into the minimum number of pieces, up to 62' wide, as required to fit the facility. Lap joints with a minimum joint width of 1/2-inch shall be used. After fabrication, the lining shall be accordion folded in both directions and packaged for minimum handling in the field. Wooden boxes for packaging shall be used for all linings which weigh over 600 pounds to prevent damage to the lining during shipment.

O4 - PLACING OF PVC LINING

- A. General. The PVC lining shall be placed over the prepared surfaces to be lined in such a manner as to assure minimum handling. It shall be sealed to all concrete structures and other openings through the lining in accordance with details shown on drawings submitted by the contractor and approved by the Engineer. The lining shall be closely fitted and sealed around inlets, outlets, and other projections through the lining. Any portion of lining damaged during installation by any cause shall be removed or repaired by using an additional piece of lining as specified hereinafter.
 - 1. Field Joints. Iap joints shall be used to seal factory fabricated pieces of PVC together in the field. Iap joints shall be formed by lapping the edges of pieces a minimum of 2 inches. The contact surfaces of the pieces shall be wiped clean to remove all dirt, dust, moisture, or other foreign materials. Sufficient cold-applied STAFF vinyl-to-vinyl adhesive shall be applied to both contact surfaces in the joint area and the two surfaces pressed together immediately. Any wrinkles shall be smoothed out.
 - 2. Joints to Structures. All curing compounds and coatings shall be completely removed from the joint area. Joining of PVC to concrete shall be made with STAFF vinyl-to-concrete adhesive. Unless otherwise shown on the drawings, the minimum width of concrete shelf provided for the cemented joint shall be 8 inches.
 - 3. Repairs to FVC. Any necessary repairs to the PVC shall be patched with the lining material itself and cold-applied STAFF vinyl-to-vinyladhesive. The adhesive shall be applied to the contact surfaces of both the patch and lining to be repaired and the two surfaces pressed together immediately. Any wrinkles shall be smoothed out.
 - 4. Quality of Workmanship. All joints, on completion of the work, shall be tightly bonded. Any lining surface showing injury due to scuffing, penetration by foreign objects, or distress from rough subgrade shall, as directed by the Engineer, be replaced or covered and sealed with an additional layer of PVC of the proper size.

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STAFF INDUSTRIES INC. 1982 West Jefferson Avenue • Detroit, Michigan 48216 • Telephone (313) 259-1820

August 14, 1975

Beker Industries Corporation P.O. Box 37 Conda, Idaho 83230

Attention: Bob Cook, Engineering Supervisor

Subject: Conditioning of reservoir for liner installation.

Dear Mr. Cook:

This will confirm our telephone conversation of August 13, in which, we discussed the preconditioning of the reservoir floor and dyke surfaces for installation of the liner which we are furnishing.

While the puncture resistance of Polyvinyl Chloride liner membrane is probably the best of all the widely used liner materials, it is still essential for satisfactory installation of this seepage barrier that the surface on which the liner is laid be a smooth well compacted surface free of all foreign objects which may project into the liner surface. It is recommended, if the soil is stony or coarse in texture that a layer of sand be used under the liner. You indicated that the surface in contact with the liner would be gpysum which could be worked to a fine grain size. We suggested that this surface be rolled immediately prior to the spreading of the liner thus minimizing surface irregularities caused by tires, running water, etc. This proceedure has worked very well on prior occasions and tends to minimize the labor involved by eliminating duplication of effort in maintaining a surface condition over a period of weeks of reservoir construction.

As regards to those areas (5 to 7 acres) which you indicated were soft due to an unlayment of wet clay, it will be essential that the surface be such to support a vehicle required to transport 5000 lb. pallets of the liner material for strategic placement and unfolding. The floor material will be furnished in pieces 100 ft. wide x 640 ft. long which will require transporting across the width of the reservoir in increments of 100 ft.

It is essential that the floor subsurface be such as to preclude any sink hole formations, which will stretch the liner beyond its elongation capabilities.



STAFF INDUSTRIES INC. 1982 West Jefferson Avenue • Detroit, Michigan 48216 • Telephone (313) 259-1820

While the steepness of the slopes on the perimeter dykes make the compacting of the side slopes difficult, it will be essential that they be graded as smooth as possible and raked free of surface irregularities. The reinforced material on the slopes is more tolerant of surface condition and we can minimize actual walking on the slopes prior to and during liner installation.

Our installation supervisor will contact you a few days prior to the liner installation work and will review the site for incidental details of site preparation.

We look forward to working with you and to a successful liner installation.

Very truly yours, Staff Industries, Inc.

Shields

cc: CE Staff



STÅFF INDUSTRIES INC. 240 Chene Street · Detroit, Michigan 48207 · Telephone (313) 259-1820

October 5, 1976

Mr. David Best Beker Industries, Corporation P. O. Box 37 Conda, Idaho 83280

Dear Mr. Best:

We are in receipt of your blueprint drawing 09-1-27 on your Decant System. You are right, you do have a little problem. Just off hand, is there any reason why the whole box can't be lowered the depth of the outlet pipe? This would make a flush bottom and would obviously eliminate some problems. At the ditch, I don't know if you have enough elevation to allow this.

Then, I would use 30 mil PVC around the redwood box and go out a few feet from the base where I would join it to the floor PVC.

Also, the Decant is about three feet too high for the Berm. If you made it the way the drawing shows you will have the same problem you had last spring, namely waves washing over the Berm and destroying it.

As you possibly know, we have four engineers at Staff Industries, but, none are practicing civil engineers. I would like to suggest that you contact a civil engineering firm that specializes in this type of installation.

Namely:

Hovater-Way Engineers, Inc. 23011 Moulton Parkway Suite F-5 Laguna Hills, California 92653

There are other details which should be considered and I highly recommend you contact the above engineers.

If we can be of further service, please advise.

Sincerely yours,

STAFF INDUSTRIES, INC.

Edward H. Staff

c.c. CE Staff EHS/njs

Sales Office: Post Office Box 797 · Upper Montclair, New Jersey 07043 · Telephone (201) 744-5367

[(v)]

STAFF INDUSTRIL : INC. 1982 West Jefferson Avenue • Detroit, Michigan 48216 • Telephone (313) 259-1820

September 19, 1975

Boker Industries, Corp. P.O. Box 37 Conda, Idaho 83230

Subject: P.O. COF 20841

Contlemen:

This is to acknowledge that Staff Industries has completed satisfactorily the installation of the liner for your No.3 Gypsum Fond at Conda, Idaho in accordance with the terms of the above subject purchase order.

It is understood that payment for this work is due and payable to our supervisor upon signing this acknowledgement in accordance with Staff Industries acceptance of your purchase order.

The Guarantees per subject purchase order are still in effect as stated in purchase order.

Staff Industries, Inc.

Date signed:

Beker Industries

Title BY: Mylen & Kely

Director
Environmental Servi

October 14, 1975 Environmental Services

Title





To: Mr. Leo Hilinski

Copies To:

M. Green

K. Rowles

File Reference:

Date: March 12, 1976

From: Gerald L. Sneddor

Department:

General Manager

Location:

Charles F. White Plant

Subject:

PRODUCT LOSS FROM NO. 3 GYPUM POND

Enclosed is our estimate of direct production loss attributable to the failure of the no. 3 Gypsum Pond.

GLS/sr

enc.

Lloyd Hemman

File Reference:

Capies To. E. M. Green Don Baumgartner ✓ G. Sneddon

March 6, 1976 Date:

From:

Gary L. Dahms &

Department:

Acid

Location:

C. P. White Plant

Conda, Idaho

GYP POND DIKE FAILURE

Pollowing is my estimate of losses resulting from the failure in the Gyp Pond Dike on March 2, 1976.

The total amount of water lost was 367 acre feet or 120,000,000 gallons. By breaking the north dike of the pond, 86 acre feet (28,000,000 gallons) were diverted to the #3 tailings pond. This leaves 281 acre feet (92,000,000 gallons) which flowed south from the pond.

Using an analysis of 0.8% P_2O_5 , 3,990 tons of P_2O_5 were lost from the pond.

GLD/kjp

-,	ĩc:	Lloyd Hemman	File Relerance:	•
Copies	To:	E. M. Green B. Cole	Date:	March 9, 1976
• •		E. Christiansen G. Sneddon	From:	Gary Dahms glong Dha
			Gepartment:	Acid
	•	10	· Location:	C. F. White Plant Conda, Idaho

Sabject: ACID AREA PRODUCTION LOSSES

The acid area production losses resulting from the #3 gypsum tailings pond dike failure were:

•	Phosphori	c Acid Plant	Sulfuric Acid	
Date	Downtime	Production Loss	Plant Production Loss	
3/2/76	16	485 Tons .	198	
3/3/76	24	735 Tons	249	
3/4/76	18.25	608 Tons	260	
3/5/76	6.75	276. Tons	256	
3/6/76	•		252	
Total	65 Hours	2,104 Tons P ₂ 0 ₅	1,215 Tons H ₂ SO ₄	

During this downtime, washed out portions of the gypsum tailings line were replaced and the line was routed back to the #2 pond. Washington Construction was utilized to prepare the #2 pond for service. The total phosphoric acid area downtime, including startup problems related to the two-day pond outage was 65 hours. Based on Pebruary's daily production rate of 735 TPD P205 the total production loss was 2,104 tons of P205. This total includes losses due to reduced production rates during the startup period thru March 5, 1976.

The sulfuric acid plants were not shut down during this time. However, with no return condensate from the phosphoric acid plant, sulfuric acid production rates had to be reduced to coincide with boiler feed water availability. Based on the 1700 TPD sulfuric acid production rate prior to the pond failure, 1,215 tons of sulfuric acid were lost between 3/2/76 and 3/6/76.

GD/kjp

Gary Dahms

L. Hemman

M. Green/

D. Baumgartner

G. Sneddon

March 10, 1976

Cal Whiting

Expansion

C. F. White Plant Conda, Idaho

LOG OF EVENTS REGARDING GYP POND DIKE FAILURE

Following is a schedule of events for 3/1/76 thru 3/2/76:

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Louisian

March 1, 1976 - Snowing and Cold - Temperature: H +230 L +100

- 1. Washington was working on the dike build up at No. 3 pond and ramp to cooling pond.
- 2. It was too stormy so Washington sent most of their crew home. We asked them to leave a cat and grader to keep the dike road open.

March 2, 1976 - Cold and Clear - Temperature: H +26° L +5°

- 1. I made rounds at 8:00 A.M. and found the dike leaking pretty bad at the drain outlet. The flume was washed out and had dropped down from gradual slope from outlet pipe approximately 15' 0.
- 2. The outlet pipe was suspended partially in mid air from center of the dike because of washout from leak under the dike.
- 3. I have no idea how much water was running thru, around, or under the culvert.
- 4. I asked Washington to start pushing fill into the dike since they were there and had already seen the situation.
- 5. I immediately notified the following people: Gary Dahms, Lloyd Hemman, and Moody Green of the situation.
- 6. Washington Construction was contacted to haul oversized rock (4" and larger) to the job site from the north end. Their trucks would not start so I notified Industrial of the urgency of hauling rock with their smaller Mack trucks (20 ton carry units).
- 7. Two D9 cats pushed fill and two 20 ton haul trucks hauled rock for 5½ hours. We tried plywood sheets and plastic along with the gyp fill. We would just about have the leak blocked and then it would break loose from under-cutting deeper in the dike. At about 1:30 P.M. as I was directing a load of rock to be dumped, the bank gave away 2' behind me. An area approximately 15' wide and 15' deep broke loose. There was not much we could do at this point.

Gary Dahms March 10, 1976 Page Two

- 8. I asked about letting loose of the dike at the north side which would allow drainage into the tailings pond. At approximately 3:00 P.M. I waded thru the water across from the south calciner and up to the area where Washington had a D9 cat and operator still working. He and I went to the north dike and began cutting the dike at 4:15 P.M. We had the water running by the time another cat came in from the north end of the plant from Washington Construction.
- 9. At approximately 10:00 P.M. the south dike was being sealed in (the pressure was off this area by now). The north end was plugged at approximately 9:00 A.M. the following morning. It was cold all night at a -16° .
- 10. I began working on the plant access road on March 3, 1976. The ponds were being taken care of by operations and Washington Construction.

CW/kjp

Affiliated FM Insurance

New Providence Corporation, Underwriting Manager

LOSS REPORT

DIFFERENCE IN CONDITIONS - MISCELLANEOUS PERILS

FORBEKER INSUDIRIES CORPORATION Conda Rd. & Rte. 34 N Soda Springs, Idaho INDEX 79228.61A

ACCOUNT 23-28879
DATE OF LOSS March 3, 1976

DATE INSPECTED June 7,1976 June 8,1976 LOSS NO. NPC-01220

BYR.K. Okimote

WITH See Below

Conference With Messrs:

J. Sneddon, Gen. Mgr; K.R. Rowles, Pers. Mgr.;

D. Best, Proj. Engr; C. Whiting, Proj. Eng;

G.L. Dahms, Acid Area Supt.

SUMMARY

The dike of Gypsum Pond No. 3 broke at the drain outlet: at this Agricultural Chemical Plant. About 92 million gals. of reclaim water dumped onto nearby farm land.

DESCRIPTION

Gypsum Pond No. 1 is a cooling pond where clear water is pumped back into the plant to reclaim the phosphoric acid still in the water. Gypsum Pond No. 2 was not in service. Gypsum Pond No. 3 is divided into two sections. The south section is the Gypsum settling pond and contains about 92 million gals. of water with an 8% concentration of phosphoric acid P_2O_5 . The north section is a tailings pond.

INCIDENT

Workmen, building up the dike on Gypsum Pond No. 3 noticed no sign of leakage Monday afternoon. However, Tuesday, March 2, they noticed leakage about 7:15 a.m. at Gypsum Pond No. 3 south section drain outlet and notified superiors who arrived at the leak about 7:45 a.m. The 36 in. diameter drain outlet pipe (reinforced fiberglass) was suspended partially in mid air due to severe wash out under the pipe by a stream of water. Two bulldozers pushed Gypsum and two 20 ton haul trucks dumped rock 4 in. and larger for about 5½ hours (8:00 a.m. - 1:30 p.m.) to block the leak. Plywood sheets, plastic liner, and other fill materials were also used but could not stop the leak, about 1:30 p.m., a 15 ft. wide by 15 ft. deep section broke loose releasing the contents of the pond.

Prepared by Factory Mutual Engineering Association

This report is intended to assist you in reducing the possibility of loss to the property insured by the Affiliated F M Insurance Company by bringing to your attention liazards and lack of protection which need promot consideration to prevent such loss to property. It is not intended to imply that all other hazards and conditions are under control at the time of this inspection. The liability of the Affiliated F M Insurance Company is limited to that covered by its insurance policies. No other liability is assumed by recsun of this report as it is only advisory in nature and the final decisions must be made by you.

DAMAGE AND SALVAGE

Ninety-two million gals. of water with an 8% concentration of phosphoric acid were lost. Six electric motors for the transfer pumps had to be rewound. The north dike of Gypsum Pond No. 3 (south section) was cut open to allow drainage into the north section. The hole caused by the leak in the dike was 20 to 30 ft. wide at the top and 20 to 21 ft. deep. The overflow pipe structure and concrete flume were washed out. Gypsum Pond No. 3 is being repaired and will be returned to service.

INTERRUPTION TO PRODUCTION

The two Sulfurice Acid Plants and the Phosphoric Acid Plant were shut down for about 65 hours, March 2 to 6, 1976 due to the time required to clean out the No. 2 pond and install the proper outlet and inlet lines from the acid plants, work was slow due to below freezing weather.

CAUSE

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The failure was caused by a leakage around the drain outlet which quickly eroded away the Gypsum fill material and grew more quickly than could be filled.

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DISCUSSION

The overflow system installed in Gypsum Pond No. 2 was modified from the system used for Gypsum Pond No. 3 and uses a wood overflow box, which is a three sided box with on side open. This box can be adjusted for overflow rate and the level of water maintained above the deposited Gypsum by adding or removing boards on the open side of the box. (See sketch). The "old" system (installed Sept. 1975) for Gypsum Pond No. 3 used more of a dam principle and since the pipe was about 15 ft. higher than the level of the deposited Gypsum, a minium of 15 ft. water level had to be maintained, whereas with the overflow box the level of the water can be kept at 1 ft... above the deposited Gypsum. Using this box system the exposure to leakage is greatly reduced because the amount of water is greatly reduced.

RECOMMENDATIONS

- 1. The new overflow system should be designed to minimize the chances of leakage around the area of the new drain pipe.
- 2. The repair work of Gypsum Pond No. 3 (south) should be expedited as much as practical so business interruption will be minimized should Gypsum Pond No. 2 develop a leak.

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NOTE: Bottom of pand is 10 mil Pic Present (or 3-4 H, above pipe) plastic liner & Side. lovel of roadway in ADRIL, 76 (a 2-31% above Pipe) walls have Bmil polyester reinfaced Bain reinf. Fiberglass drain outlet coated pue liner. _Brucen flume (concreb) Despos tea Initial wash out area Gilber (How Situation Appeared at Bios A.M. toes. March 2,1976)
Per description & statch by C. WHITING, Phys. Engr. SIDE YIEW or cross section looking east. TOP of PLAN. YIEM (as described by G. L. DAHINS, Acid Supl.) NOTE: NO DRAWINGS OF THE PIPE SUPPORTS F Details of liner seal at drain could be found at the plant. spaning to box is adjusted by adding boords as layers (3 sides closed) Deposition core, flume (シーリンエットッ General features of Proposed new overflow System presently in planning Stage. INDUSTRIES CORPORATION SODA SPRINGS, IDAHD For her Egal, of E. K. Chimoto Scale: No SCALE Ind. No. 7922 8.61 FACTORY MUTUAL ENGINEERING ASSOCIATION Factory Mutual System Tr. No.





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70 Alex Carrozelli

Jack Asche

Copies To:

G. Sneddon

D. Dean

L. Hemman

D. Fleming

D. Best

File Reference:

Date:

December 28, 1976

From:

Garv L. Dahms

Department:

Phos Acid Administration

Location:

C. F. White Plant Conda. Idaho

Subject: #3 GYPSUM TAILINGS POND DIKE FAILURE

The acid loss claimed in the #3 Gypsum Tailings Pond Dike Failure was all water soluble, recoverable product. This means that the acid was basically in a useable form. Although it was not concentrated enough to use at this point, all of the raw material costs to make product acid had been put into it. It need only be recycled through the digestion reaction in place of the raw water required for the reaction to be converted to $28\%~P_2O_5$ acid. The only costs to make this conversion would be the pumping costs which would be more than offset by the pumping costs of the well water which is used when pond water is not available.

The volume of solution used in the claim was based only on the liquid which was "free-standing" above the gypsum. Although there was a substantial amount of water in the gypsum, none of this was used because of the difficulty in calculating the amount and the strength of this acid. The solution used in the claim was only that which was already separated from the gypsum solids.

There has been speculation by outside groups that some P_2O_5 is lost from standing in the ponds due to chemical reactions and settling with the gyp. Since we are dealing with the solution which was already separated from the gypsum, there is no basis for the belief that some of it may have settled in with the gypsum tailings. This may occur, but we are not working with the solution in the gypsum. It was already separated from the gypsum.

Our analysis of the gypsum tailings pond system indicated there would be no loss of P_2O_5 due to chemical reactions in the "free-standing" portion of the pond. The pH of the pond water is 1.0 to 1.2 at a temperature of $32^{\circ}C$. At these conditions, all P_2O_5 would exist as H_3PO_4 and the ions necessary to form iron and aluminum phosphate would not be present. The P_2O_5 concentration is less than 1%. With this temperature and P_2O_5 strength the dihydrate form of gypsum is extremely stable (See Figure 5). The solubility of this gypsum in the pond water would be 0.25% (See Figure 1). The free sulfuric acid in the pond is 0.55%. This low concentration of gypsum and relatively high free sulfuric acid would prohibit the loss of any P_2O_5 in the form of dicalcium phosphate in the "free-standing" portion of the pond. Another possible P_2O_5 loss could result from substitution of P_2O_5 in the gypsum crystal lattice which could bind P_2O_5 and precipitate it with the gypsum. As can be seen in Figure 2, the pond is somewhat out of the range at which this can occur (100 G/L = 9% P_2O_5). In addition, the "free-standing" portion of this pond would not have enough gypsum present to promote this reaction.

A certain amount of P_2O_5 upgrading occurs in the pond as it cools. The average temperature of the pond is $43^{\circ}C$ at the inlet and $30^{\circ}C$ at the outlet. The inlet solution is saturated with sodium and potassium fluosilicates. During the

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cooling process, the solubility of sodium fluosilicate drops from 1.3% to 0.7%. The solubility of potassium fluosilicate drops from 0.6% to 0.4% (See Figure 3). This loss of diluents would result in a 0.8% upgrading of the P_2O_5 in solution. Even more upgrading would result from the normal evaporation rate in this area.

The P_2O_5 in the pond water is recovered by using it for the water of hydration in the formation of dehydrate gypsum and for the dilution water in the product. Some of this water is added directly to the digestion reaction. The remainder is first used for the three stage countercurrent wash on the filters and then added to the digester. When the water is used in the reaction the P_2O_5 content reports to the 28% P_2O_5 product acid with no further treatment.

A P_2O_5 balance in the pond system is shown in Figures 6,7, and 8. Figure 6 shows a balanced pond system with the phos acid plant operating at 93% total P_2O_5 recovery. Well water is added to the digester. All of the filter wash water comes from the cooling pond. Figure 7 shows the pond system while the #3 pond was filling up and the water could not be recovered. In this case the sluice water had to be raw water and the amount of wash water was somewhat reduced. Additional well water had to be added to the digestion system to balance the reaction. Figure 8 shows the system which is used when the pond level gets high. This is the system shich would have been used to reclaim the pond water. In this system, very little well water is added to the digestion system. The P_2O_5 which is in the pond water added to the digester reports directly to the product stream and moves on to the evaporation step. This is not a hypothetical case. It has been used in the past to reduce pond levels. When this case is in practice, the actual plant recoveries are 94.9%.

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Alex Carrozelli

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Date:

December 28, 1976

From:

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Department:

Phos Acid Administration

Location:

C. F. White Plant

Conda, Idaho

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The volume of solution used in the claim was based only on the liquid which was "free-standing" above the gypsum. Although there was a substantial amount of water in the gypsum, none of this was used because of the difficulty in calculating the amount and the strength of this acid. The solution used in the claim was only that which was already separated from the gypsum solids.

There has been speculation by outside groups that some P205 is lost from standing in the ponds due to chemical reactions and settling with the gyp. Since we are dealing with the solution which was already separated from the gypsum, there is no basis for the belief that some of it may have settled in with the gypsum tailings. This may occur, but we are not working with the solution in the gypsum. It was already separated from the gypsum.

Our analysis of the gypsum tailings pond system indicated there would be no loss of P205 due to chemical reactions in the "free-standing" portion of the pond. The pH of the pond water is 1.0 to 1.2 at a temperature of 32°C. At these conditions, all P_2O_5 would exist as H_3PO_4 and the ions necessary to form iron and aluminum phosphate would not be present. The P205 concentration is less than 1%. With this temperature and P_2O_5 strength the dihydrate form of gypsum is extremely stable (See Figure 5). The solubility of this gypsum in the pond water would be 0.25% (See Figure 1). The free sulfuric acid in the pond is 0.55%. This low concentration of gypsum and relatively high free sulfuric acid would prohibit the loss of any P205 in the form of dicalcium phosphate in the "free-standing" portion of the pond. Another possible P205 loss could result from substitution of P205 in the gypsum crystal lattice which could bind $P_2^{\,0}$ and precipitate it with the gypsum. As can be seen in Figure 2, the pond is somewhat out of the range at which this can occur (100 G/L = 9% P_2O_5). In addition, the "free-standing" portion of this pond would not have enough gypsum present to promote this reaction.

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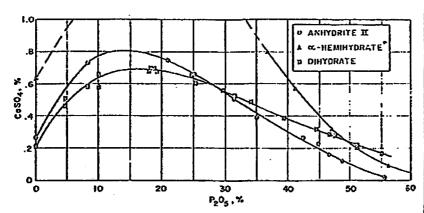


FIG. 9. Solubility of dihydrate, a-hemihydrate*, and anhydrite II in phosphoric acid solutions at 25°C. [After Taperova and Shulgina (6).]

Figure 1

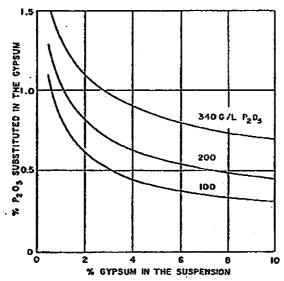
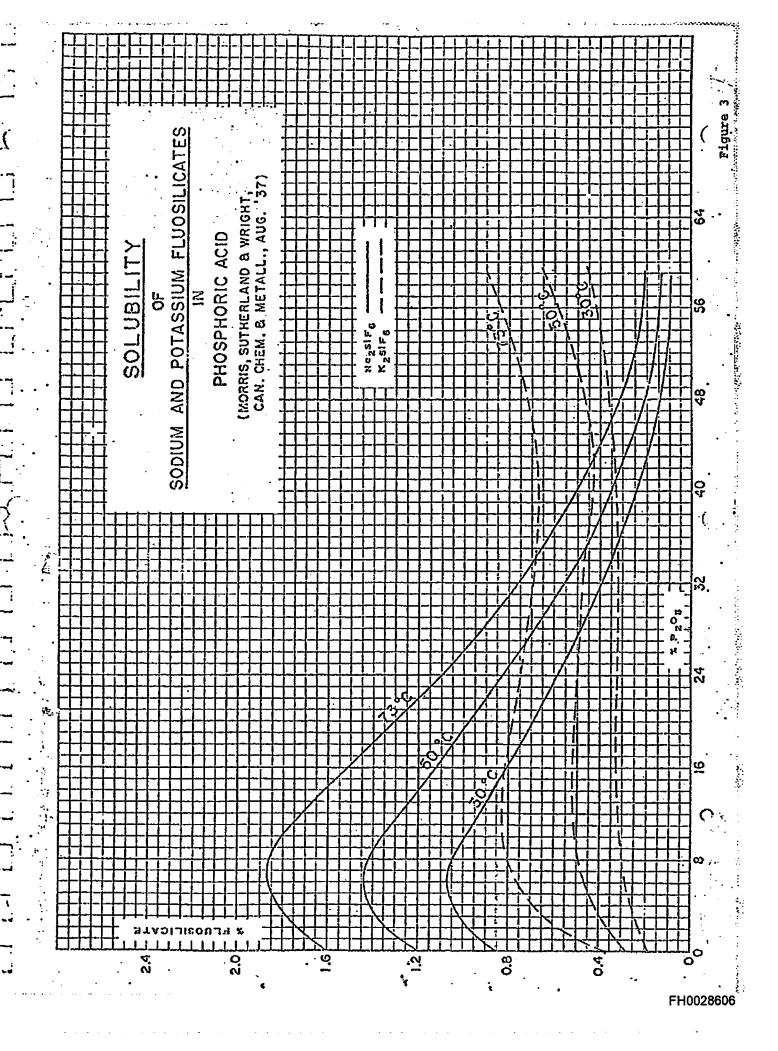
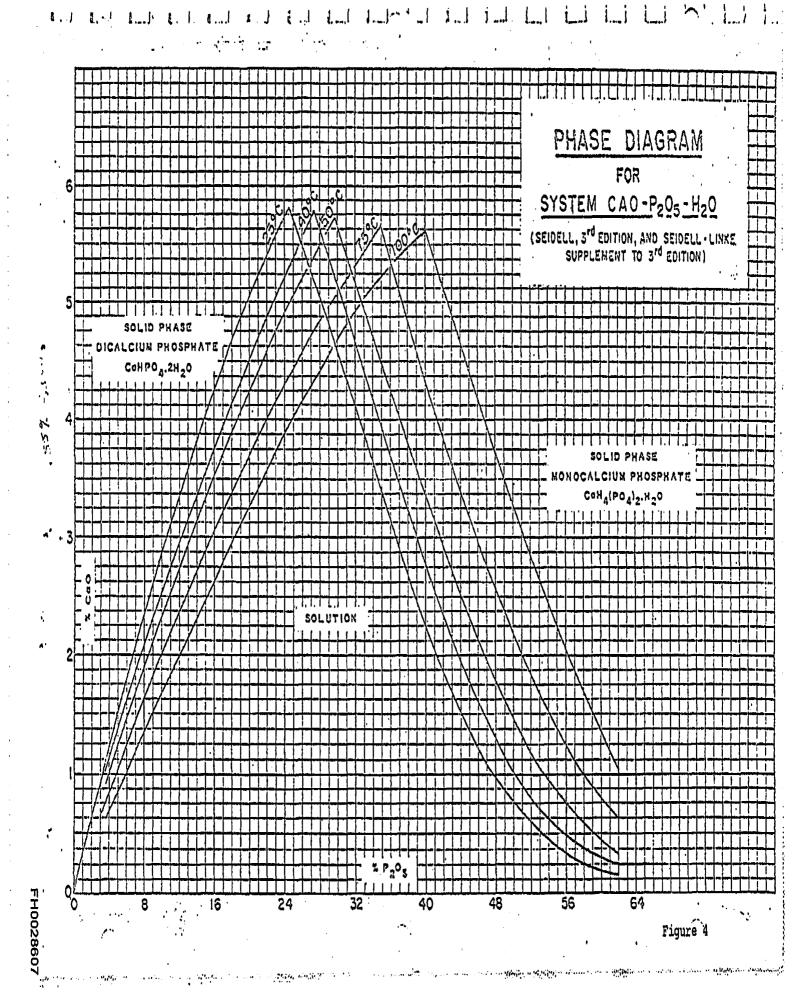
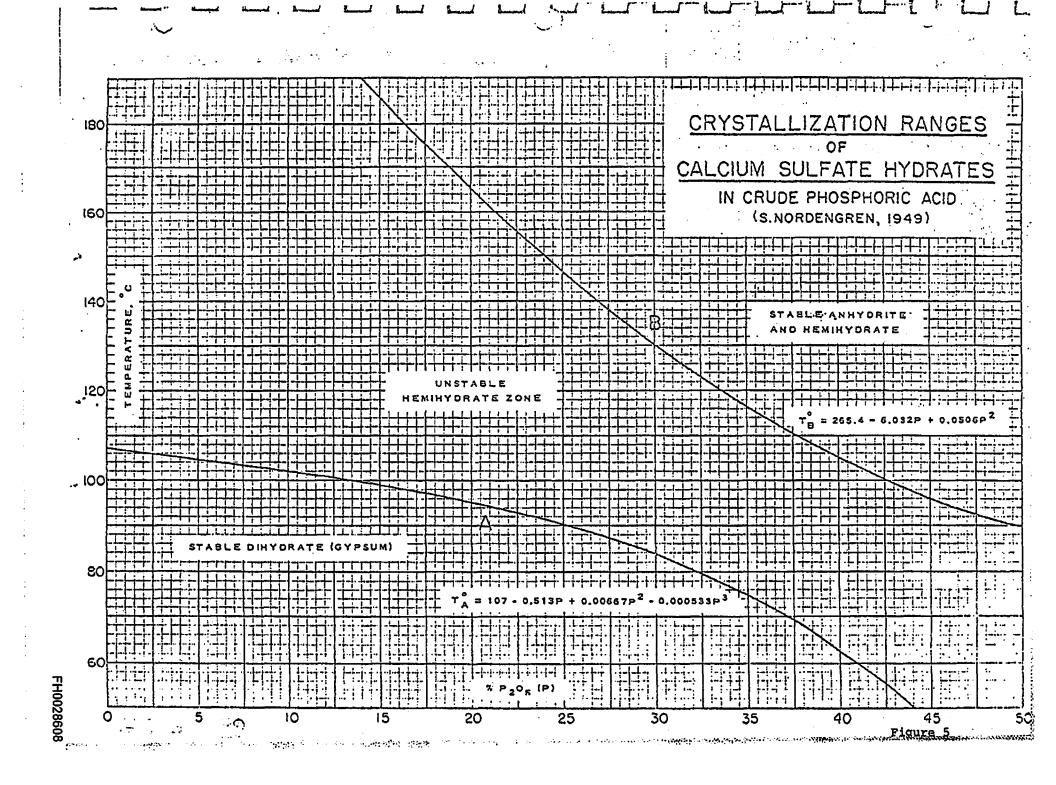
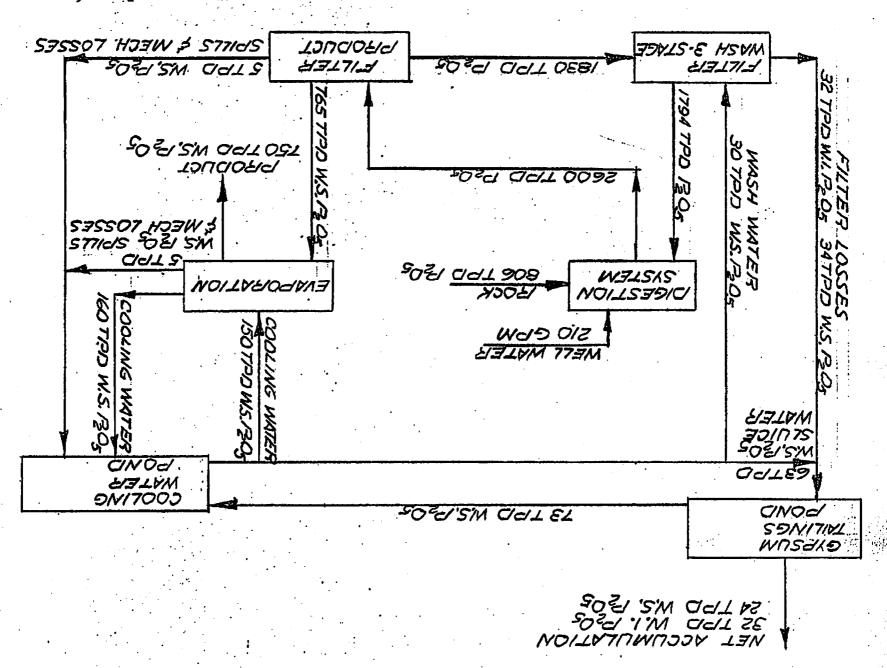


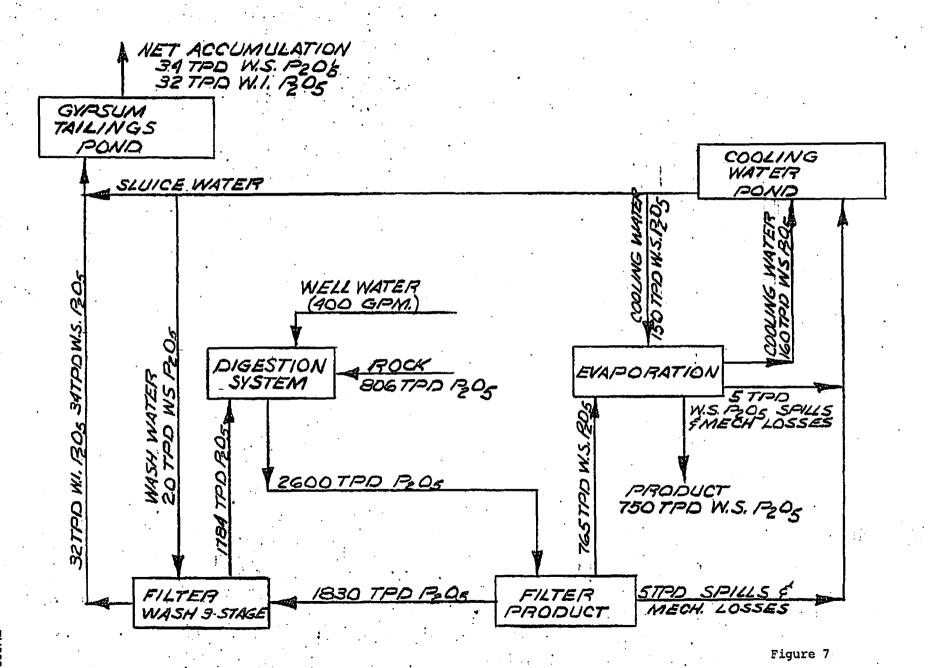
FIG. 8. Substitution of P₂O₂ in gypsum at 70°C as a function of the percentage of solids in suspension and the momentary concentration of P₂O₂ in the mother Figure 2 liquor. [After Früchen and Becker (23).]



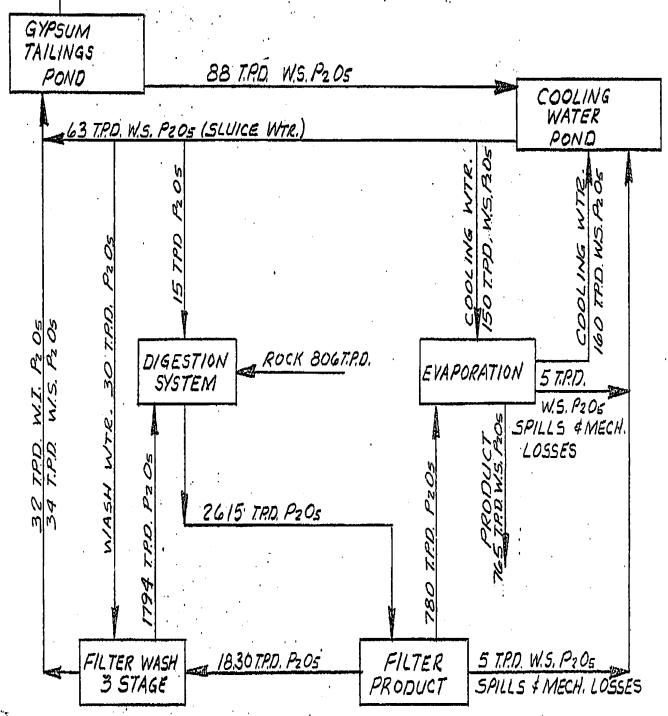








CASE III RECLAIMING P205 NET ACCUMULATION IN GYP TAILINGS POND 9 T.P.D. W.S. P20s 32 T.P.D. W.I. P20s



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